Three New ent-Labdane Diterpenoids from the Wood of Excoecaria agallocha LINN.

by Ponnapalli Mangala Gowri^{*a}), Sri Vedavyasa Srirangaraja Radhakrishnan Bhattar^a), Poreddy Guruva Reddy^a), Yerraballi Rakesh^a), Shaik Jeelani Basha^b), AkellaVenkata Subrahmanya Sarma^b), and Janaswamy Madhusudana Rao^a)

 ^a) Natural Products Laboratory, Indian Institute of Chemical Technology, 500007 Hyderabad, India (phone: +91-40-27191735; fax: +91-40-27160512; e-mail: mangala@iict.res.in)
^b) NMR Division, Indian Institute of Chemical Technology, 500007 Hyderabad, India

An extensive study of metabolites present in *Excoecaria agallocha* LINN. led to the isolation of three new *ent*-labdane-type diterpenoids, named agallochaexcoerins A-C (1-3), besides three known compounds. The skeleton present in compound 1 is rather unusual, containing of a seven-membered lactone. The structures were elucidated on the basis of spectroscopic studies and comparison with known related compounds. The isolated compounds 1-6 were not active against Raw 264.7 (macrophage-like), K 562 (leukemia), and COLO 205 (colon) human carcinoma cell lines.

Introduction. - The wide-spread genus Excoecaria of Euphorbiaceae comprises 40 species which are distributed on seashores, edge mangroves throughout tropical countries including Africa, Asia, Northwest Australia, and India [1]. It is well known that the milky latex exuded from the bark of this plant is poisonous and may cause temporary blindness and blistering of the skin [2]. In traditional Thai medicine, the bark and wood of the plant are used to combat flatulence. Several skin irritant daphnane and tigliane diterpene esters have been isolated from the latex of E. agallocha [3][4]. Recently, a new phorbol ester acting as an anti-HIV agent, was isolated from the stems and leaves of this species collected in Northwest Australia [5]. Furthermore, the anti-tumor activities of some of these diterpenes were investigated [6] [7]. More recently, a series of diterpenoid derivatives with different kinds of carbon skeletons, such as labdane, beyerane, isopimarane, and kaurane types, were isolated from this plant [6][8-11]. We report herein the structure determination of three new diterpenoids, agallochaexcoerin A, a novel seco-labdanoid with a rare seven-membered lactone ring, and agallochaexcoerins B and C through spectral data and also by chemical correlation.

Results and Discussion. – The analysis of an acetone extract of the powdered wood of *E. agallocha* led to the isolation of three new *ent*-labdane diterpenoids 1-3. Three known compounds were also identified by comparison of their spectral data with those reported in the literature as ribenone **4** [12], *ent*-8,13-epoxy-11 α -hydroxy-3-oxo-13-*epi*-labd-14-ene **5** [12], and octacosyl (*E*)-ferulate **6** [13].

Compound 1 was obtained as colorless needles. The HR-ESI-MS of 1 exhibited the *pseudo*-molecular-ion peak at m/z 359.2183 ($[M + Na]^+$), which established the

^{© 2009} Verlag Helvetica Chimica Acta AG, Zürich



molecular formula as $C_{20}H_{32}O_4$, indicating five degrees of unsaturation. The IR spectrum showed absorption bands ascribable to OH (3377 cm^{-1}) , vinyl (1625, 949, 921 cm⁻¹), CO (1722, 1290 cm⁻¹), and ether (1136 cm⁻¹) functionalities. It was recognized as a new 13-epi-8,13-epoxy-seco-labdanoid by interpretation of the ¹Hand ¹³C-NMR spectral data (Tables 1 and 2). The presence of a vinyl group was evident from the spectra. The presence of five Me groups (*Tables 1* and 2) of an intact labdane skeleton suggested that it might be a ring A-seco-derivative. The appearance of signals for geminal Me(18) and Me(19)¹), deshielded at $\delta(H)$ 1.34 and 1.27, suggested the presence of tertiary OH group at C(4), i.e., a 3,4-seco-labdanoid with the possibility of C(3) being part of an acid or a lactone. The location of the tertiary OH group at C(4)was also supported by the HMBC correlation between C(4) (δ (C) 75.8) and Me(18), Me(19), and H–C(5). The ¹³C-NMR spectrum showed two O-bearing C-atoms with signals at $\delta(C)$ 72.9 (s) and at 72.1 (s) accounting for C(8) and C(13) of a 13-epimanoyloxy partial structure [14], and two more O-bearing C-atoms with signals at $\delta(C)$ 70.9 (d) and at 75.8 (s). Compound 1 didn't react with diazomethane, suggesting the presence of a lactone, rather than of a carboxylic acid. The lactone could, therefore, be between C(3) and C(11), or C(3) and C(6). If the O-bearing CH group was at C(6), the ¹³C-NMR signal for C(7) would be expected around δ (C) 50–53. However, if the Obearing CH group was at C(11), the signal for C(12) would be around δ (C) 40-45. The

¹⁾ Arbitrary numbering. For systematic names, see Exper. Part.

	1 ¹) ^a)	2 ^a)	3 ^b)	3a ^b)
1α	1.52 - 1.62 (m)	2.10 (d, J = 12.2)	2.40 (d, J = 12.9)	2.48 $(d, J = 13.9)$
1β	3.04 (dd, J = 8.0, 13.2)	2.47 (d, J = 12.2)	3.23 (d, J = 12.9)	-
2α	2.47 (dd, J = 7.3, 13.2)	-	-	-
2β	2.76(t, J = 13.2)	-	-	-
3		3.88 (br. $d, J = 1.0$)	3.91 (d, J = 3.6)	4.93 (br. s)
5	1.38 - 1.43 (m)	1.62°)	1.68 (dd, J = 2.6, 13.3)	-
6α	1.65 - 1.72 (m)	1.81 ^d)	1.81 (qd, J = 3.4, 13.3)	-
6β	1.43 - 1.49 (m)	1.40 - 1.44 (m)	1.42 (dq, J = 3.0, 13.3)	-
7α	1.43 - 1.49 (m)	1.56 ^e)	1.54 (dt, J = 3.4, 13.3)	-
7β	1.81 (dd, J = 3.6, 9.5)	1.92 (td, J = 3.4, 12.5)	1.86 (td, J = 3.4, 13.3)	-
9	1.96 (d, J = 8.0)	1.64 ^c)	1.60 (d, J = 9.4)	-
11α	4.76 (dd, J = 7.3)	1.45 - 1.51 (m)	-	-
11β	-	1.56 ^e)	4.16 (td, J = 4.6, 9.4)	5.18 - 5.22 (m)
12α	2.05 (d, J = 15.4)	1.67 (dd, J = 6.4, 8.5)	-	-
12β	2.39 (dd, J = 6.6, 15.4)	1.81 ^d)	2.46 (dd, J = 4.6, 13.5)	2.40 (d, J = 13.9)
14	5.88 (dd, J = 10.7, 17.3)	5.87 (dd, J = 10.7, 17.3)	5.97 (dd, J = 12.9, 10.9)	5.90 (dd, J = 10.8, 17.5)
15a	5.18 (dd, J = 1.3, 17.3)	4.94 (dd, J = 1.5, 10.7)	4.96(d, J = 11.1)	4.98(d, J = 11.2)
15b	4.97 (dd, J = 1.3, 10.7)	5.16 (dd, J = 1.5, 17.3)	5.07 (d, J = 17.7)	5.23 (d, J = 17.9)
16	1.42 (s)	1.30 (s)	1.25 (s)	1.27(s)
17	1.19 (s)	0.77 (br. s)	0.88(s)	0.86(s)
18	1.34 (s)	1.18 (s)	1.19 (s)	1.11 (s)
19	1.27 (s)	0.69(s)	0.68(s)	0.82(s)
20	1.24 (s)	1.30 (s)	1.23 (s)	1.25 (s)
OH	-	3.43 (br. s)	3.48 (d, J = 4.9)	-
^a) M	easured at 500 MHz. b) I	Measured at 600 MHz. °)	, ^d), ^e) Overlapping sign	als.

Table 1. ¹H-NMR Data of 1, 2, 3, and 3a

C-atoms C(7) and C(12) appeared at δ (C) 42.9 (t) and 37.6 (t), respectively, to support a lactone bridge between C(3) and C(11). The CO group at δ (C) 175.9 (s) showed HMBC correlations with H-C(11), H-C(1), and H-C(2), supporting the presence of a seven-membered lactone ring between C(3) and C(11). The lactone was found to be a secondary lactone with the O-bearing CH group resonating at $\delta(C)$ 70.9 (d) and the corresponding CH group at $\delta(H)$ 4.76 (dd, J = 7.3) (coupling with H–C(9) and one of the vicinal H-atoms at C(12); the other H-atom at C(12) did presumably not show a coupling due to a dihedral angle of ca. 90°, with ring C in a quasi-chair form). A literature survey revealed that the first member of a 2,3-seco-labdanoid with δ -lactone between C(2) and C(11), agallochin N (7), has recently been reported from the same species of Indian origin [9]. The axial disposition of the O-bearing CH group at C(11) was comparable to agallochin N with an ¹H-NMR signal for the a 11β -oxymethine group at $\delta(H)$ 4.63 (td, J=12.0, 4.0), while the corresponding signals for 13-epimanoyloxide derivatives containing an 11α -OH group were noticed at 4.18 [14], 4.13 [15], and 4.17 ppm [12]. This is in accordance with the fact that an equatorial H-atom in a tetrahydropyran ring is more deshielded than its axial counterpart, as described in [16].

Agallochaexcoerin A thus appears to be the first example of a *seco*-tricyclic labdanoid containing a rare seven-membered lactone group between C(3) and $C(11)^1$).

Table 2.	$^{13}C-NMR$	Spectral	Data o	f 1.	2.	3.	and	3a
I uoio 2.	0 1 1 1 1 1 1	Specular	Dana O	/ = 4	_	· • •	001000	~~~

	1 ¹) ^a)	2 ^a)	3 ^b)	3a ^b)
$CH_{2}(1)$	39.1	52.1	54.6	54.5
$CH_2(2)$	29.7	210.9	211.8	203.7
C(3)	175.9	82.9	82.3	83.7
C(4)	75.8	45.2	44.8	42.7
H-C(5)	58.3	54.4	54.4	55.2
$CH_2(6)$	23.6	19.6	19.3	19.3
$CH_2(7)$	42.9	42.6	42.7	42.9
C(8)	72.9	74.6	76.5	75.7
H-C(9)	56.7	55.3	62.2	59.1
C(10)	41.5	43.1	43.8	39.7
$H-C(11)$ or $CH_2(11)$	70.9	15.6	64.9	67.7
CH ₂ (12)	37.6	35.5	44.9	39.4
C(13)	72.1	73.5	74.1	73.6
H - C(14)	146.8	147.5	147.2	146.4
CH ₂ (15)	110.9	110.6	110.2	111.1
Me(16)	29.5	28.3	31.9	31.9
Me(17)	16.7	16.3	16.2	17.1
Me(18)	34.9	29.3	29.4	28.9
Me(19)	27.8	16.2	17.3	17.2
Me(20)	25.8	24.9	25.1	25.6
MeO-CO-C(3)				170.5
MeO-CO-C(3)				20.6
MeO-CO-C(11)				170.1
MeO-CO-C(11)				21.7

The ¹³C-NMR spectral data of agallochaexcoerin A can be compared with a 3,4-*seco*labdanoid (8), which was isolated from the same species, as well as synthesized from ribenone with *m*-CPBA in 7.6% yield along with (*R*)- and (*S*)-epoxides (20 and 25%, resp.) [17]. Unlike in 1, compound 8 doesn't possess a OH group at C(11), and therefore no lactone between C(3) and C(11).

Agallochin N [9], agallochin E (8,13-epoxy-3-nor-2,3-seco-13-epi-labden-2,4-olide) [8], 2-oxo-3-oxamanoyl oxide [18], excoecarin H [7], ent-13-epi-8,13-epoxy-2-oxa-3oxolabd-14-ene-(1*R*)-carboxylic acid [10], and excoecarin M [19] with each containing a δ -lactone moiety, were reported from the same species. The NOESY correlations of H-C(5) and H-C(9), Me(17) and Me(20), H-C(14) and Me(20), and H-C(11) and Me(20) established the relative configuration. Agallochaexcoerin A was considered as an ent-derivative in view of its laevo specific rotation [10][20] to derive its structure as ent-8,13-epoxy-4-hydroxy-3,4-seco-13-epi-labd-14-en-3,11-olide.

Compound **2** was isolated as colorless needles from aqueous EtOH. ESI-MS of **2** exhibited a *pseudo*-molecular-ion peak at m/z 355 ($[M + Cl]^+$), ascribable to a molecular formula $C_{20}H_{32}O_3$. The presence of OH (3440 cm⁻¹), CO (1714 cm⁻¹), mono-substituted C=C bond (1660, 1451, 1014, 918 cm⁻¹), and ether (1160 cm⁻¹) functionalities were noticed in the IR spectrum. A preliminary analysis of the ¹H- and ¹³C-NMR spectral data (*Tables 1* and 2) suggested it to be a new 8,13-epoxy-13-epi-

labdane diterpenoid. The presence of a vinyl and a keto group were evident from the spectra. The ¹³C-NMR spectrum showed an O-bearing C-atom at $\delta(C)$ 82.9 (d), the CO group at $\delta(C)$ 210.9 accounted for the 2-keto group in addition to two O-bearing Catoms at 74.6 (s), 73.5 (s), accounting for the 8,13-epoxymanoyloxy structure [8]. The presence of an isolated CH₂ group in α -position to a keto group was indicated by an AB system at $\delta(H)$ 2.10 and 2.47. The Me(18) and Me(19) attached to C(4) with resonances at $\delta(H)$ 1.18 and 0.69, respectively, indicated the presence of an α -ketol system when compared to the signals of those of 2-oxomanoyl oxide ($\delta(H)$ 0.87 and 1.08) [21]. The location of the keto CO group at C(2) was also supported by HMBC (Table 3 and Fig. 1) data, where the C(2) showed correlation with H_{ax} -C(1), H_{eq} -C(1), H_{ax} -C(3), Me(18), and Me(19). The NOESY correlations between H-C(5) and H-C(9), Me(17) and Me(20), H-C(14) and Me(20), and H-C(6) and Me(20) established the relative configuration of 2, as shown in Fig. 2. It was related to the epi-manoyl oxide skeleton by the correlations found in the ¹H,¹H-NOESY data (*Table 3* and *Fig. 2*). A search in the literature revealed that the isomeric manoyl oxide 9 has been isolated before from the plant Lagarostrobos colensoi (Dacrydium colensoi) [21] and Euphorbia segetalis [22]. A comparative study of 2 and 9 revealed that they have identical ¹H- and ¹³C-NMR spectra, just like manoyl oxide and 13-epi-manoyl oxide, but that they differ in their physical properties, such as m.p. and CD. The circular dichroism (CD) spectrum of 2 showed a negative Cotton effect at 299 nm, analogous to that of **4**. The absolute configuration of **2** was assumed to be that of an *ent*-derivative in the light of its laevo specific rotation. An inspection of the vast literature [23] revealed



Fig. 1. Key HMBC and COSY correlations of agallochaexcoecarins A, B, and C (1-3)



Fig. 2. NOESY Correlations of agallochaexcoecarins A, B, and C (1-3)

			Table 3. <i>HMBC</i>	COS.	Y, and NOESY (Correlations	of Compound.	s 1, 2 , a	nd 3		
	1 ¹)				2				3		
	HMBC	COSY	NOESY		HMBC	COSY	NOESY		HMBC	COSY	NOESY
1α	3, 17	$1\beta, 2\beta$	17	1α	2, 9, 10, 17	1β	I	1α	2, 3, 10, 17	1β	17
1β	2, 3, 9, 10	$1\alpha, 2\alpha$	I	1β	2, 3, 5, 10, 17	1α	19	1β	2, 9, 10, 17	1α	I
2α	3, 10	$1\beta, 2\beta$	$11\alpha, 17$	3α	2, 4, 18, 19	I	$1\alpha, 12\alpha, 18$	3α	2, 4, 18, 19	I	18
2β	1, 3, 10	$1\alpha, 2\alpha$	I	5β	I	I	I	5β	4, 17, 19	6β	I
5α	4	9	9α	6α	I	I	9α	6α	I	6β	I
6α	I	7	I	6β	I	I	I	6β	I	6a	I
6β	I	7	I	7α	I	7β	I	7α	I	7β	I
7α	I	6	9α	$_{7\beta}$	I	7α	I	$_{7\beta}$	I	7α	I
Ţβ	I	I	I	θ_{β}	I	I	I	6	8, 10, 11, 17, 20	I	I
9α	8, 10, 11, 17, 20	11α	5α , 7α	11α	I	I	5α	11β	I	12α	$12\alpha, 17, 20$
11α	3, 10, 12, 13	$12\beta, 9\alpha$	2β , 12β , 17 , 20	11β	I	I	I	12α	9, 11, 13	$11\beta, 12\beta$	11β
12α	9, 11, 16	$12\beta, 11\alpha$	I	12α	I	$11\alpha, 12\beta$	I	12β	11, 13	12α	20
12β	13, 14, 16	12α	$11\alpha, 20$	12β	I	12α	I	14	I	15a, 15b	15a, 20
14	13	15a, 15b	15a, 20	14	I	15a, 15b	15a, 20	15a	13, 14	14, 15b	14
15a	13, 14	14, 15b	I	15a	I	14, 15b	14	15b	13	14, 15a	I
15b	13	14, 15a	I	15b	I	14, 15a	I	16	12, 13	I	I
16	12, 13, 14	I	I	16	12, 13, 14	I	I	17	1, 5, 9, 10	I	$1\alpha, 11\beta$
17	1, 5, 9, 10	I	$1\beta, 2\beta, 11\beta, 20$	17	1, 5, 9, 10	I	I	18	3, 4, 5, 19	I	3a
18	4, 5, 19	I	I	18	2, 3, 4, 5, 19	I	I	19	3, 4, 5, 18	I	I
19	4, 5, 18	I	I	19	2, 3, 4, 5, 18	I	I	20	7, 8, 9	I	$11\beta, 12\beta$
20	7, 8, 9	I	$11, 12\beta, 14, 17$	20	7, 8, 9	I	I				

Helvetica Chimica Acta – Vol. 92 (2009)

that *ent*-labdanes are laevo-rotatory, while labdanes are dextro-rotatory [24]. Therefore, the absolute configuration of agallochaexcoerin B was tentatively fixed as *ent*-8,13-epoxy- 3β -hydroxy-13-*epi*-labd-14-en-2-one.

Compound **3** was obtained as colorless needles. The ESI-MS of **3** displayed a *pseudo*-molecular-ion peak at m/z 359 ($[M + Na]^+$), corresponding to a molecular formula of C₂₀H₃₂O₄. The IR spectrum of **3** revealed the presence of OH (3443 cm⁻¹), CO (1716 cm⁻¹), olefinic (1661, 1453, 1114, 919 cm⁻¹), and ether (1160 cm⁻¹) groups. On acetylation with pyridine and acetic anhydride, **3** gave a diacetate **3a**. The molecular formula C₂₄H₃₆O₆ of **3a** was deduced from *quasi*-molecular-ion peak, at m/z 443.2407 ($[M + Na]^+$) in the HR-ESI-MS. The IR spectrum of **3a** showed acetate (1735, 1225 cm⁻¹) and ether (1125 cm⁻¹) moieties, but no OH absorption, suggesting that **3** has two acylable OH groups but no free tertiary OH group, which might be present as an ether.

The ¹H-NMR spectrum of **3** was similar to that of **2**, except for the presence of an extra signal for an O-bearing CH group at $\delta(H)$ 4.16, and three Me singlets at $\delta(H)$ 1.25, 1.23, and 0.88, which were shifted upfield by 0.05, 0.07, and downfield by 0.11 ppm, respectively, compared with the corresponding Me signals of 2. Comparison of the ¹³C-NMR spectra (*Table 2*) of **3** and **2** suggested that the OH group was located at the C(11), based on the substitution effects at the respective α and β positions. The configuration at C(11) in **3** was assigned from the ${}^{1}H$, ${}^{1}H$ coupling constant and ${}^{13}C$ chemical shifts. The signal of H-C(11) at $\delta(H)$ 4.16 (dt, J = 4.6, 9.4) was identical to the corresponding signal for ent-11a-hydroxy-3-oxo-13-epi-manoyl oxide [12]. Further, the corresponding signal in the diacetate at $\delta(H)$ 5.18–5.22 (m), was deshielded. A transdiaxial relationship between the two H-atoms H-C(9) and H-C(11) was assumed, based on the coupling constant between these two H-atoms. Therefore, the orientation of the OH group at C(11) was presumed to be equatorial. The relative configuration was deduced by the NOESY spectrum (Fig. 2, Table 3). The absolute configuration of 3 was considered as an *ent*-derivative in view of its laevo-specific rotation to derive the structure as ent- 3β ,11 α -dihydroxy-2-oxo-13-epi-manoyl oxide. The CD spectrum of **3** showed a negative *Cotton* effect at 288 nm analogous to that of **4**. Therefore, the absolute configuration of **3** was tentatively fixed as *ent*-8,13-epoxy- 3β ,11 α -dihydroxy-13-epi-labd-14-en-2-one.

In cytotoxic studies, compounds 1-6 were tested against human carcinoma cell lines using the MTT (=(3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) assay. However, none of the compounds showed statistically significant cytotoxicity ($IC_{50} > 50 \mu g/ml$) for the cell lines tested.

Experimental Part

General. Column chromatography (CC): silica gel (SiO₂; 100–200 or 230–400 mesh, *Acme*). Analytical TLC: Silica gel 60 F 254 plates (0.25 mm, *Merck*). M.p.: *Fischer–John micro* melting point apparatus; uncorrected. Optical rotations: *Horiba DIP-370* digital polarimeter. CD Spectra: *Jasco J-810* spectrometer. IR Spectra: *Thermo Nicolet Nexus* 670 spectrometer; with KBr pellets. ¹H-NMR (300, 500, and 600 MHz) and ¹³C-NMR (75, 100, and 150 MHz) spectra: *Bruker Avance-300, Varian INOVA-400, Varian INOVA-500*, or *Bruker Avance-600* spectrometer in CDCl₃ with TMS as internal standard; coupling constants (*J*) in Hz. MS: *JEOL MS-BU 20* or a *JEOL LMS-SX-120A QQ* mass spectrometer.

Plant Material. The wood of *Excoecaria agallocha* (1.5 kg) was collected from Visakhapatnam coast (latitude 17° 42′ 0″ N, longitude 83° 18′ 0″ E), Andhra Pradesh, India in August 2007, and was identified by Assoc. Prof. Dr. *Venkiah*, Dept. of Botany, Andhra University, Visakhapatnam. A voucher specimen (IIC-101) has been deposited with the Herbarium of the Indian Institute of Chemical Technology.

Extraction and Isolation. The chopped resinous wood (1.2 kg) was exhaustively extracted three times with acetone in a *Soxhlet* apparatus (3×51). Removal of the solvent from the combined acetone extracts gave a brown syrup (25 g). A portion (22 g) of this brown syrup was subjected to CC over SiO₂ using solvent mixtures of increasing polarity from hexane through acetone to yield several fractions. *Fr.* 5 (500 mg) was chromatographed on SiO₂ with 5% acetone to afford ribenone (**4**; 10 mg), 6% acetone to afford agallochaexcoerin C (**3**; 6 mg), and 7% acetone to afford octacosyl (*E*)-ferulate (**6**; 5 mg). *Fr.* 10 (100 mg) was subjected to SiO₂ CC using 10% acetone to afford *ent*-11*a*-hydroxy-3-oxo-13-*epi*-manoyl oxide (**5**; 3 mg), 12% acetone to afford agallochaexcoerin B (**2**; 5 mg), and 14% acetone to afford agallochaexcoerin A (**1**; 10 mg).

Agallochaexcoerin A (=(2R,3aS,7aR,8S,10aS,10bR)-2-Ethenyldecahydro-8-(2-hydroxypropan-2-yl)-2,7a,10a-trimethyloxepino[2,3,4-de]chromen-5(2H)-one; **1**). Colorless needles. M.p. 106–108°. [α]₂₇²⁷ = -32.0 (c = 1.0, CHCl₃). UV (EtOH): 230 (4.03), 320 (4.05). IR (KBr): 3377, 1722, 1625, 1290, 1136, 949, 921. ¹H- and ¹³C-NMR: *Tables 1* and 2. HR-ESI-MS: 359.2183 ([M + Na]⁺, C₂₀H₃₂NaO₄⁺; calc. 359.2192).

Agallochaexcoerin B (=(3R,4aS,6aS,8R,10aR,10bS)-3-Ethenyldodecahydro-8-hydroxy-3,4a,7,7,10apentamethyl-9H-benzo[f]chromen-9-one; **2**). Colorless needles. M.p. 72–74°. [a] $_{27}^{27}$ = -59.9 (c = 1.0, CHCl₃). CD (2 × 10⁻⁴, MeOH): 0 (325), -0.47 × 10⁻⁴ (299), 0 (243). IR (KBr): 3440, 1660, 1451, 1160, 1014, 918. ¹H- and ¹³C-NMR spectra: *Tables 1* and 2. ESI-MS: 355 ([M + Cl]⁺).

Agallochaexcoerin C (=(1R,3R,4aS,6aS,8R,10aR,10bR)-3-Ethenyldodecahydro-1,8-dihydroxy-3,4a,7,7,10a-pentamethyl-9H-benzo[f]chromen-9-one; **3**). Colorless needles. M.p. 128–130°. [a] $_{27}^{27}$ = -69.9 (c = 1.0, CHCl₃). CD (2×10^{-4} , MeOH): 0 (329), -0.45×10^{-4} (288), 0 (239). IR (KBr): 3443, 1716, 1661, 1453, 1160, 1114, 919. ¹H- and ¹³C-NMR spectra: Tables 1 and 2. ESI-MS: 359 ([M + Na]⁺).

 3β ,11 α -Diacetoxyagallochaexcoerin C (=(1R,3R,4aS,6aS,8R,10aR,10bR)-3-Ethenyldodecahydro-3,4a,7,7,10a-pentamethyl-9-oxo-1H-benzo[f]chromene-1,8-diyl Diacetate; **3a**). Colorless needles. ¹Hand ¹³C-NMR: Tables 1 and 2. HR-ESI-MS: 443.2407 ([M + Na]⁺, C₂₄H₃₆NaO⁺₄; calc. 443.2409).

Ribenone (=(3R,4aS,6aS,10aR,10bS)-3-*Ethenyldecahydro-3,4a,7,7,10a-pentamethyl-1*H-*benzo[f]-chromen-8(4a*H)-one; **4**). White needles. M.p. 109–110°. $[\alpha]_{27}^{27} = -53.0$ (c = 1.0, CHCl₃). IR (KBr): 2941, 2858, 1705, 1458, 1383, 1082, 1007. ¹H- and ¹³C-NMR: *Tables 1* and 2. ESI-MS: 327 ($[M + Na]^+$).

ent-11 α -Hydroxy-3-oxo-13-epi-manoyl Oxide (=(1R,3R,4aS,6aS,10aR,10bR)-3-Ethenyldecahydro-1-hydroxy-3,4a,7,7,10a-pentamethyl-1H-benzo[f]chromen-8(4aH)-one; **5**). Colorless needles. M.p. 114– 116°. [α]₂₇^T = -59.2 (c = 1.0, CHCl₃). IR (KBr): 3450, 1699, 1677, 1412, 1130, 960, 912. ¹H- and ¹³C-NMR: Tables 1 and 2. ESI-MS: 321 ([M + Na]⁺).

Octacosyl (E)-*Ferulate* (= *Octacosyl* (2E)-3-(4-Hydroxy-3-methoxyphenyl)prop-2-enoate; **6**). White needles. M.p. 78–80°. IR (KBr): 3325, 2941, 2858, 1742, 1625, 1383, 1225, 1082, 1007. ESI-MS: 587 ([M + H]⁺).

We are indebted to the Department of Science and Technology (Project GAP- 0141) and CSIR for financial support. The authors also thank Dr. J. S. Yadav, Director, I. I. C. T. for his constant encouragement.

REFERENCES

- M. Hotta, K. Ogata, A. Nitta, K. Hosikawa, M. Yanagi, K. Yamazaki, 'Useful Plants of the World', Heibonsha, Tokyo, 1989.
- [2] S. Kumarasinghe, R. Seneviratne, Aust. J. Dermatol. 1998, 39, 275.
- [3] C. Karalai, P. Wiriyachitra, H. J. Opferkuch, E. Hecker, Planta Med. 1994, 60, 351.
- [4] P. Wiriyachitra, H. Hajiwangoh, P. Boonton, W. Adolf, H. J. Opferkuch, E. Hecker, *Planta Med.* 1985, 51, 368.

- [5] K. L. Erickson, J. A. Beutler, J. H. Cardellina II, J. B. McMahon, D. J. Newman, M. R. Boyd, J. Nat. Prod. 1995, 58, 769.
- [6] T. Konoshima, T. Konishi, M. Takasaki, K. Yamazoe, H. Tokuda, Biol. Pharm. Bull. 2001, 24, 1440.
- [7] T. Konishi, T. Konoshima, Y. Fujiwara, S. Kiyosawa, Chem. Pharm. Bull. 1998, 46, 721.
- [8] A. S. R. Anjaneyulu, V. L. Rao, Phytochemistry 2000, 55, 891.
- [9] A. S. R. Anjaneyulu, V. L. Rao, Phytochemistry 2003, 62, 585.
- [10] T. Konishi, Y. Fujiwara, T. Konoshima, S. Kiyosawa, Chem. Pharm. Bull. 1998, 46, 1393.
- [11] T. Konishi, K. Yamazoe, T. Konoshima, T. Maoka, Y. Fujiwara, K. Miyahara, J. Nat. Prod. 2003, 66, 108.
- [12] T. Konishi, M. Azuma, R. Itoga, S. Kiyosawa, Y. Fujiwara, Y. Shimada, Chem. Pharm. Bull. 1996, 44, 229.
- [13] A. E. Nkengfack, D. R. Sanson, M. S. Tempesta, Z. T. Fomum, J. Nat. Prod. 1989, 52, 320.
- [14] L. Zhou, E. R. Fuentes, J. J. Hoffmann, B. N. Timmermann, Phytochemistry 1995, 40, 1201.
- [15] B. M. Fraga, P. González, R. Guillermo, M. G. Hernández, J. Rovirosa, *Phytochemistry* 1989, 28, 1851.
- [16] J. De Pascual Teresa, A. San Feliciano, Y. M. Miguel del Corral, Farm. Nueva 1976, 41, 343; Chem. Abstr. 1977, 86, 2994c.
- [17] T. Konishi, M. Takasaki, H. Tokuda, S. Kiyosawa, T. Konoshima, Biol. Pharm. Bull. 1998, 21, 993.
- [18] P. K. Grant, M. H. G. Munro, N. R. Hill, J. Chem. Soc. 1965, 3846.
- [19] T. Konishi, T. Konoshima, T. Maoka, Y. Fujiwara, Tetrahedron Lett. 2000, 41, 3419.
- [20] S. Habtemariam, A. I. Gray, C. Lavaud, G. Massiot, B. W. Skelton, P. G. Waterman, A. H. White, J. Chem. Soc., Perkin Trans. 1 1991, 893.
- [21] R. C. Cambie, S. C. Moratti, P. S. Rutledge, P. D. Woodgate, Aust. J. Chem. 1990, 43, 791.
- [22] J. Jakupovic, F. Jeske, T. Morgenstern, F. Tsichritzis, J. A. Marco, W. Berendsohn, *Phytochemistry* 1998, 47, 1583.
- [23] J. Buckingham, 'Dictionary of Chemical Compounds', 5th Edn. and supplements, Chapman and Hall, London, 1982–1987.
- [24] R. Caputo, L. Mangoni, P. Monaco, L. Previtera, Phytochemistry 1974, 13, 471.

Received December 16, 2008